



Texture and orientation characteristics of α - Al_2O_3 films prepared by laser chemical vapor deposition using Nd:YAG laser

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ABSTRACT

α - Al_2O_3 films were prepared by laser chemical vapor deposition (LCVD) and the effects of precursor vaporization temperature (T_{vap}), total chamber pressure (P_{tot}), laser power (P_L) and deposition temperature (T_{dep}) on the phase, orientation and texture of Al_2O_3 film were investigated. At $P_{\text{tot}} = 0.93$ kPa, α - Al_2O_3 films were obtained in the region of $T_{\text{vap}} > 423$ K and $T_{\text{dep}} > 1100$ K. The orientation of α - Al_2O_3 film changed from (1 1 0) to (0 1 2) to (1 0 4) to (0 0 6) with increasing P_{tot} . Porous α - Al_2O_3 films were formed at high T_{vap} (443 K) and low P_{tot} (0.47 kPa). At $T_{\text{vap}} = 413$ K, α - Al_2O_3 film had hexagonal and rectangular plate-like grains with finely faceted edges. With increasing $P_{\text{tot}} = 0.93$ –1.4 kPa, (0 0 6)-oriented α - Al_2O_3 film with a hexagonal terrace texture was obtained.

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1. Introduction

Alumina (Al_2O_3) and titanium carbonitride (Ti(C, N)) are widely used as hard-coating materials. Since Al_2O_3 is characterized by high hardness and low thermal conductivity at high temperatures, the cutting efficiency and lifetime of cemented carbide (WC-Co) tools with an Al_2O_3 coating can be improved [1]. Moreover, since WC-Co tools should be able to withstand high temperatures and harsh stress conditions, when the dry cutting process with higher speed is employed, a high-performance hard coating is strongly demanded [2].

A low temperature deposition technique with phase control is required in Al_2O_3 coating due to the many polymorphs of Al_2O_3 [3]. In particular, among these polymorphs, corundum structure (α -phase) appears to be the only thermodynamically stable phase with excellent hardness at high temperatures. Although α - Al_2O_3 is widely used for hard-coating material [4], the deposition temperature of α - Al_2O_3 coating by conventional thermal chemical vapor deposition (CVD) is about 1273 K. Such high temperature restricts the selection of substrate to only a few materials with a high thermal stability. Metastable phases such as the γ - and κ -phases would have to be prepared at lower

deposition temperatures. However, they may transform to the α -phase above 1000 K, resulting in volume change and cracking or delamination of the Al_2O_3 coating during the cutting process [5,6].

Several characteristics of α - Al_2O_3 coating such as hardness and roughness could depend on the crystallographic orientation and texture. Rупpi et al. prepared (0 1 2), (1 0 4), (0 0 1)-oriented α - Al_2O_3 films on Ti(C, O) and Ti(C, N) intermediate layers deposited on WC-Co substrate and reported that the films with (0 0 6) and (1 0 4) orientations showed higher hardness and modulus than those with random or the other orientations [4,7–9]. It is commonly understood that the crystal orientation depends not only on the substrate but also on the deposition temperature and the total pressure in the CVD process. Since α - Al_2O_3 film in a single phase can only be prepared at a high deposition temperature in a limited range by conventional thermal CVD, the relationship among deposition conditions, orientation and texture has not been studied in a wide range of conditions [1,10]. In order to develop an α - Al_2O_3 coating with control of phase, orientation and texture, a low temperature α - Al_2O_3 coating could be promising as a high-performance coating for cutting tools.

We have studied low temperature and high-speed deposition of thick Y_2O_3 and TiO_2 films with control of their orientations by laser CVD (LCVD) using a high-power Nd:YAG laser (260 W) [11,12]. In this study, we prepared α - Al_2O_3 films by LCVD and investigated the effect of deposition conditions on the phase, orientation and texture of Al_2O_3 film.

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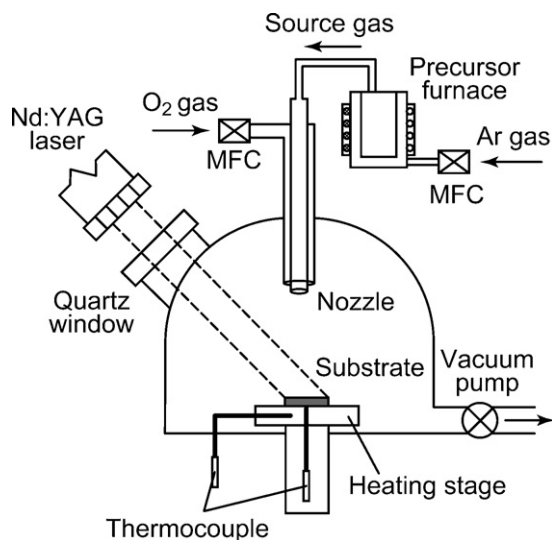


Fig. 1. Schematic of laser CVD apparatus.

2. Experimental procedure

Fig. 1 shows a schematic of the LCVD apparatus and Table 1 lists the deposition conditions. Aluminum tri-acetylacetonate ($\text{Al}(\text{acac})_3$) was used as the precursor. $\text{Al}(\text{acac})_3$ was vaporized at vaporization temperature (T_{vap}) of 453 K and carried into the CVD chamber with Ar gas. O_2 gas and $\text{Al}(\text{acac})_3$ vapor were separately introduced into the CVD chamber through a double-tube nozzle. The gas flow rates of Ar and O_2 gases were maintained at $3.3 \times 10^{-6} \text{ m}^3 \text{ s}^{-1}$ (200 sccm). The total pressure (P_{tot}) ranged from 0.47 to 2.33 kPa. Yttria-stabilized zirconia (YSZ) plates ($12.5 \text{ mm} \times 12.5 \text{ mm} \times 1.0 \text{ mm}$) were used as the substrate, and the YSZ substrate was heated on a hot stage at pre-heating temperatures (T_{pre}) from room temperature to 973 K. The distance between the nozzle and the substrate was fixed at 28 mm. An Nd:YAG laser in a continuous wave mode (wavelength: 1064 nm) was collimated by lenses to irradiate the whole substrate and introduced into the CVD chamber through a quartz window. The laser power (P_L) was changed up to 260 W. The deposition temperature (T_{dep}) was measured with a thermocouple inserted into the back side of the substrate.

The crystal phase was determined by X-ray diffraction (θ - 2θ XRD; Rigaku RAD-2C). The surface and cross-sectional texture was observed by scanning electron microscopy (SEM; Hitachi S-3100H). The deposition rate (R_{dep}) was calculated from the thickness and the deposition time.

3. Results and discussion

In common CVD process, T_{dep} is obviously one of the most dominant parameters and usually determined by heating with an electric current or by a high-frequency electromagnetic wave, etc. In the LCVD process, on the other hand, the T_{dep} can be determined for various parameters such as P_L , T_{pre} , P_{tot} and T_{vap} . For example, the P_L dependence on T_{dep} at various T_{vap} for $T_{\text{pre}} = 873 \text{ K}$ and $P_{\text{tot}} = 0.97 \text{ kPa}$ is depicted in Fig. 2. By laser irradiation, the T_{dep} increased by 150–200 K from T_{pre} , and the temperature increase from T_{pre} to T_{dep} became more than 400 K when the P_L was increased to 260 W. The temperature increase tended to be greater at higher T_{vap} . This might be caused by the increase in exothermic heat by the reaction between the precursor and O_2 gases as well as by the increase in significant plasma formation by laser irradiation of the gases.

Table 1
Deposition conditions.

$\text{Al}(\text{acac})_3$ vaporization temperature (T_{vap})	403–443 K
Substrate pre-heating temperature (T_{pre})	873 K
Gas flow rate for both Ar and O_2 gases	$3.3 \times 10^{-6} \text{ m}^3 \text{ s}^{-1}$
Total pressure (P_{tot})	0.47–2.33 kPa
Distance between nozzle and substrate	28 mm
Laser output power (P_L)	88–236 W

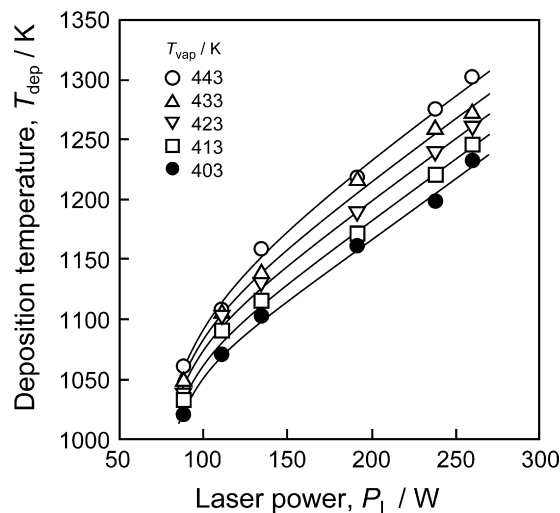


Fig. 2. P_L dependence on the T_{dep} at $T_{\text{pre}} = 873 \text{ K}$, $P_{\text{tot}} = 0.97 \text{ kPa}$ and various T_{vap} .

Fig. 3 shows the XRD patterns of Al_2O_3 films prepared at $P_L = 192 \text{ W}$. At $P_{\text{tot}} = 0.93 \text{ kPa}$ and $T_{\text{vap}} = 423 \text{ K}$ ($T_{\text{dep}} = 1210 \text{ K}$), α - Al_2O_3 film in a single phase with slight (0 1 2) and (1 0 4) orientation was obtained (Fig. 3(a)). At $T_{\text{vap}} = 443 \text{ K}$, a peak around $2\theta = 41.7^\circ$ ($d = 0.217 \text{ nm}$) from the (0 0 6) plane was identified (Fig. 3(b)), its intensity increasing at $P_{\text{tot}} = 1.4 \text{ kPa}$ (Fig. 3(c)). The diffraction intensity of the (0 0 6) plane is only 2.0 compared with 100 for the (1 0 4) plane according to the JCPDS card (No. 46-1212). Therefore, the intensity of the (0 0 6) peak cannot be identified in common α - Al_2O_3 . The Harris texture coefficient (TC) can be used to quantitatively estimate the orientation, particularly for a small diffraction

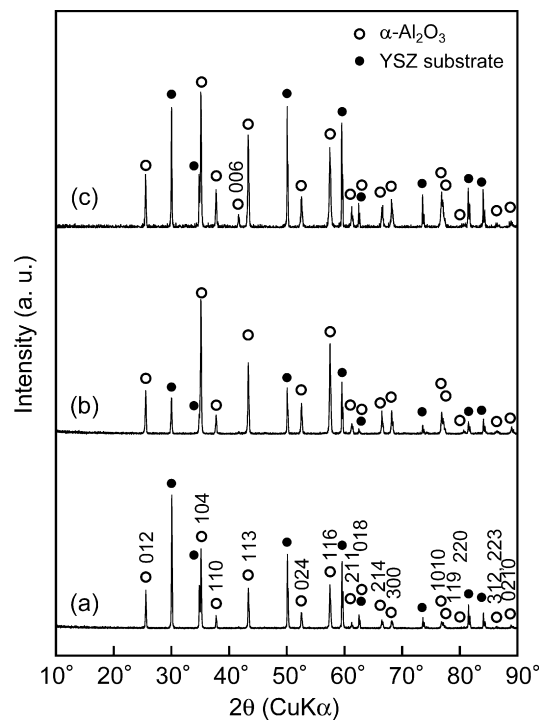


Fig. 3. XRD patterns of Al_2O_3 films prepared at $P_L = 192 \text{ W}$, $P_{\text{tot}} = 0.93 \text{ kPa}$ and $T_{\text{vap}} = 423 \text{ K}$ ($T_{\text{dep}} = 1210 \text{ K}$) (a), $P_{\text{tot}} = 0.93 \text{ kPa}$ and $T_{\text{vap}} = 443 \text{ K}$ ($T_{\text{dep}} = 1230 \text{ K}$) (b), and $P_{\text{tot}} = 1.4 \text{ kPa}$ and $T_{\text{vap}} = 443 \text{ K}$ ($T_{\text{dep}} = 1210 \text{ K}$) (c).

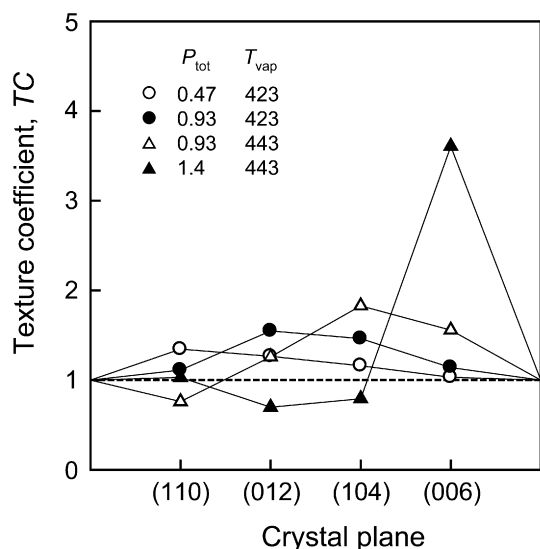


Fig. 4. Preferred orientations of Al_2O_3 films prepared at $P_L = 192$ W and various P_{tot} and T_{vap} : 0.47 kPa and 423 K (open circles), 0.93 kPa and 423 K (filled circles), 0.93 kPa and 443 K (open triangles), and 1.4 and 443 K (filled triangles).

peak [13].

$$\text{TC}(hkl) = N \frac{I_m(hkl)/I_0(hkl)}{\sum I_m(hkl)/I_0(hkl)}, \quad (1)$$

where $I_m(hkl)$ and $I_0(hkl)$ are the intensity from the (hkl) plane measured in the present study and that reported in JCPDS card (No. 46-1212), respectively. (012), (104), (110), (006), (113), (024), (116), (214) and (300) planes were used for the calculation ($N=9$). $\text{TC}(hkl)$ has a value between 0 and N depending on the orientation degree of the (hkl) plane. The TC of a non-oriented plane should be 1.0. If the TC is more than 1.0, we can define it as being oriented. If the TCs of more than two planes increase simultaneously, we can define them as being co-oriented. Fig. 4 shows the TC for (110), (012), (104) and (006) planes of $\alpha\text{-Al}_2\text{O}_3$ films prepared at $P_L = 192$ W and various P_{tot} and T_{vap} . The $\alpha\text{-Al}_2\text{O}_3$ film prepared at $P_{\text{tot}} = 0.47$ kPa and $T_{\text{vap}} = 423$ K was (110) and (012) co-oriented (open circles in Fig. 4), and that prepared at $P_{\text{tot}} = 0.93$ kPa was (012) and (104) co-oriented (filled circles in Fig. 4). At $P_{\text{tot}} = 0.93$ kPa and $T_{\text{vap}} = 443$ K, the $\alpha\text{-Al}_2\text{O}_3$ film changed to (104) and (006) co-oriented (open triangles in Fig. 4). The $\alpha\text{-Al}_2\text{O}_3$ film prepared at $T_{\text{vap}} = 443$ K and $P_{\text{tot}} = 1.4$ kPa was significantly (006) oriented (filled triangles in Fig. 4).

Fig. 5 depicts the effect of T_{dep} and T_{vap} on the phase formation of Al_2O_3 film. At $T_{\text{dep}} < 1060$ K, $\gamma\text{-Al}_2\text{O}_3$ films in a single phase were formed independently of T_{vap} . Al_2O_3 films with a $\gamma + \theta$ mixture phase were formed at $T_{\text{vap}} = 403$ K and $T_{\text{dep}} > 1070$ K, whereas Al_2O_3 films with an $\alpha + \theta$ mixture phase were formed at $T_{\text{vap}} = 413$ K and $T_{\text{dep}} > 1116$ K. At $T_{\text{vap}} > 423$ K, the phase changed from a γ to a $\gamma + \alpha$ mixture to an α -phase with increasing T_{dep} .

Fig. 6 depicts the effect of P_{tot} and T_{vap} on the phase and orientation of Al_2O_3 film. At $P_{\text{tot}} = 0.47$ kPa, $\alpha\text{-Al}_2\text{O}_3$ film was prepared at $T_{\text{vap}} > 413$ K. With increasing T_{vap} , the orientation changed from (110) to (110) and (012) co-orientation to (104) and (012) co-orientation. At $P_{\text{tot}} = 0.93$ kPa, $\alpha\text{-Al}_2\text{O}_3$ film was also prepared at $T_{\text{vap}} > 423$ K, and (012) and (104) co-oriented and (104) and (006) co-oriented $\alpha\text{-Al}_2\text{O}_3$ films were prepared. A significant (006)-oriented $\alpha\text{-Al}_2\text{O}_3$ film was obtained at $P_{\text{tot}} = 1.4$ kPa and $T_{\text{vap}} = 443$ K. At $P_{\text{tot}} > 1.9$ kPa and $T_{\text{vap}} < 423$ K, $\gamma\text{-Al}_2\text{O}_3$ film in a single phase was formed, and an α -phase appeared with a γ -phase at $T_{\text{vap}} > 433$ K. It can be summarized that at $T_{\text{vap}} > 423$ K, the Al_2O_3 film prepared at $P_{\text{tot}} < 0.93$ kPa tended to be α -phase, while the Al_2O_3 film at $P_{\text{tot}} > 1.4$ kPa tended to be γ -phase. At lower T_{vap} , a θ -phase

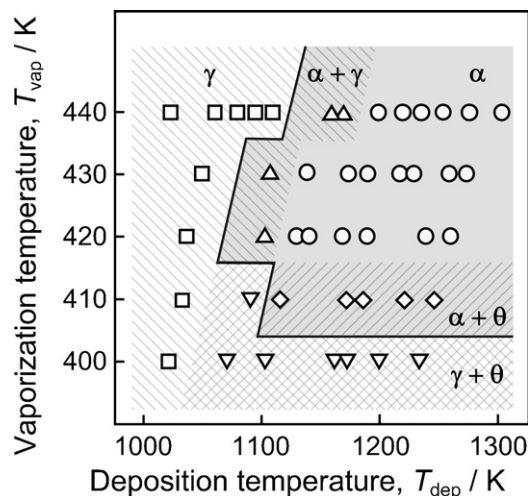


Fig. 5. Effect of T_{dep} and T_{vap} on the phase of Al_2O_3 films prepared at $P_{\text{tot}} = 0.93$ kPa.

appeared with the α - and γ -phases. The orientation of $\alpha\text{-Al}_2\text{O}_3$ film changed from (110) to (012) to (104) to (006) with increasing P_{tot} .

Fig. 7 shows the surface SEM images of Al_2O_3 films prepared at various T_{vap} , P_{tot} and P_L . (104) and (006) co-oriented $\alpha\text{-Al}_2\text{O}_3$ film prepared at $T_{\text{vap}} = 443$ K, $P_{\text{tot}} = 0.93$ kPa and $P_L = 192$ W had hexagonal faceted grains slightly inclined to the substrate (Fig. 7(a)). (006)-oriented Al_2O_3 film was obtained at $P_{\text{tot}} = 1.4$ kPa, and a hexagonal terrace texture developed on the surface of the grains (Fig. 7(b)). When the P_L increased to 238 W at $P_{\text{tot}} = 0.93$ kPa, rounded hexagonal and rectangular-shaped grains were observed (Fig. 7(c)). At $T_{\text{vap}} = 443$ K and $P_{\text{tot}} = 0.47$ kPa, porous $\alpha\text{-Al}_2\text{O}_3$ film was formed (Fig. 7(d)). As in the conventional CVD process, a porous structure was often obtained under a high supersaturation condition of source gases. In the present study, the gas supply rate at $T_{\text{vap}} = 443$ K and $P_{\text{tot}} = 0.47$ kPa increased to a level two times higher than those of the other deposition conditions. When the T_{vap} decreased to 433 K, hexagonal and rectangular plate-like grains were grown with finely faceted edges (Fig. 7(e)). At $P_{\text{tot}} = 2.3$ kPa, $\gamma\text{-Al}_2\text{O}_3$ film having a cauliflower-like structure in a single phase was obtained (Fig. 7(f)). A typical cross-sectional SEM image of oriented $\alpha\text{-Al}_2\text{O}_3$ films is shown in Fig. 8. The $\alpha\text{-Al}_2\text{O}_3$ films with (104) and (006) co-orientation prepared at $T_{\text{vap}} = 433$ K, $P_{\text{tot}} = 0.93$ kPa and $P_L = 192$ W had a columnar structure. The step structure was

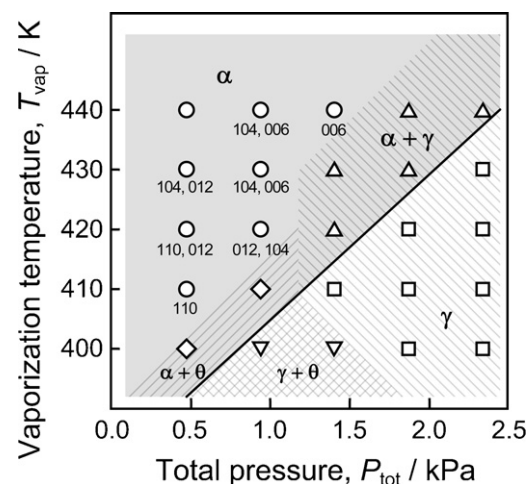


Fig. 6. Effect of P_{tot} and T_{vap} on the phase and orientation of Al_2O_3 films prepared at $P_L = 192$ W.

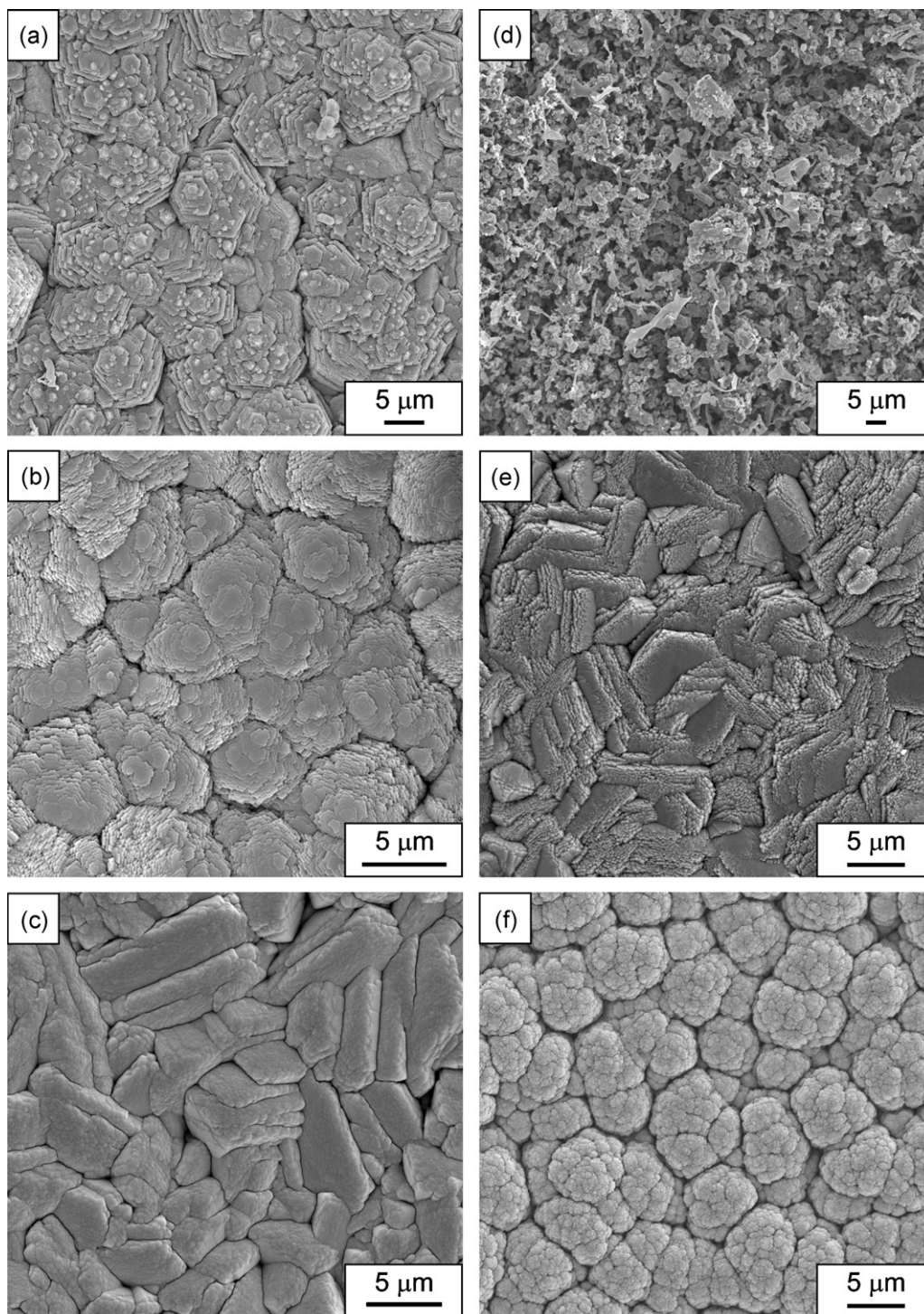


Fig. 7. Surface SEM images of Al_2O_3 films prepared at $T_{\text{vap}} = 443$ K and various P_{tot} and P_L : $P_{\text{tot}} = 0.93$ kPa and $P_L = 192$ W (a), $P_{\text{tot}} = 1.4$ kPa and $P_L = 192$ W (b), $P_{\text{tot}} = 0.93$ kPa and $P_L = 238$ W (c), and at $P_L = 192$ W and various P_{tot} and T_{vap} : $P_{\text{tot}} = 0.47$ kPa and $T_{\text{vap}} = 443$ K (d), $P_{\text{tot}} = 0.47$ kPa and $T_{\text{vap}} = 433$ K (e), and $P_{\text{tot}} = 2.3$ kPa and $T_{\text{vap}} = 433$ K (f).

observed at the surface of columnar grains and the feather-like texture was formed in the cross-section. The thickness of the most $\alpha\text{-Al}_2\text{O}_3$ films was ranging from 15 to 25 μm and the highest R_{dep} was 540 $\mu\text{m h}^{-1}$ at $T_{\text{dep}} = 1100$ K for $\gamma\text{-Al}_2\text{O}_3$ film, and was 324 $\mu\text{m h}^{-1}$ at $T_{\text{dep}} = 1200$ K for $\alpha\text{-Al}_2\text{O}_3$ film.

Fig. 9 summarizes the effect of P_{tot} and T_{vap} on the texture of Al_2O_3 films prepared at various deposition conditions. At low P_{tot} (0.47 kPa) and high T_{vap} (443 K), porous $\alpha\text{-Al}_2\text{O}_3$ film was formed. With increasing P_{tot} to 1.4 kPa, the $\alpha\text{-Al}_2\text{O}_3$ film had columnar grains with a hexagonal terrace texture, while with decreasing T_{vap}

to 413 K, $\alpha\text{-Al}_2\text{O}_3$ film had hexagonal and rectangular plate-like grains with finely faceted edges. At higher P_{tot} or lower T_{vap} , γ - and θ -phases appeared with the α -phase, and non-oriented grains were grown with a hexagonal and rectangular faceted texture. At $P_{\text{tot}} > 1.4$ kPa, $\gamma\text{-Al}_2\text{O}_3$ film with a cauliflower-like texture was obtained.

The orientation and texture of $\alpha\text{-Al}_2\text{O}_3$ film also depended on the deposition conditions such as T_{vap} and P_{tot} . Schematics of the cross-sectional crystal structure of the (006), (104), (012) and (110) planes of $\alpha\text{-Al}_2\text{O}_3$ are presented in Fig. 10. In the corundum

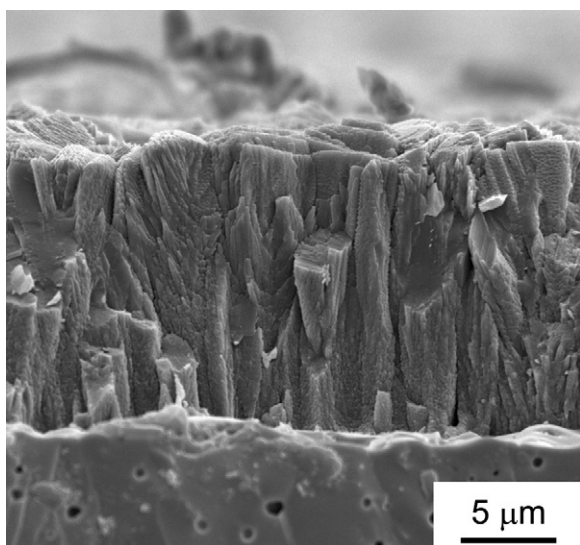


Fig. 8. Cross-sectional SEM image of α - Al_2O_3 films with (104) and (006) co-orientation prepared at $T_{\text{vap}} = 433$ K, $P_{\text{tot}} = 0.93$ kPa and $P_L = 192$ W.

structure of α - Al_2O_3 , the Al_2O_9 octahedral dimers are joined alternately by edge- and corner-sharing in the c -axis direction. (012) and (104) planes intersect with the (006) plane at angles of 57.6° (Fig. 10(b)) and 38.2° (Fig. 10(c)), respectively. The orientation of α - Al_2O_3 film changed from (110) to (012) to (104) to (006) with increasing P_{tot} , as summarized in Fig. 5. It may be assumed that the longer direction of Al_2O_9 octahedral dimers (corresponding to the packing density of O^{2-} ions) decreased with increasing P_{tot} and T_{vap} . It is generally understood that the (001) orientation of hexagonal lattice would appear at a low supersaturation condition in vapor and liquid phase deposition in various systems. In the present LCVD, (006) oriented α - Al_2O_3 films were obtained under rather high P_{tot} conditions. Since homogeneous reactions in a gas phase (mainly powder formation) occurs more significantly at higher P_{tot} , resulting in depletion of source gases, supersaturation would be decreased with increasing P_{tot} . This might have caused the (006) orientation at high P_{tot} .

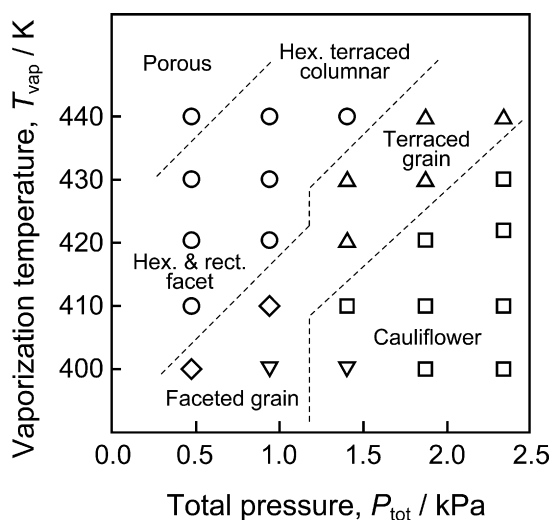


Fig. 9. Effect of P_{tot} and T_{vap} on the texture of Al_2O_3 films prepared at $P_L = 192$ W. Plot shapes correspond to the phase (Fig. 5).

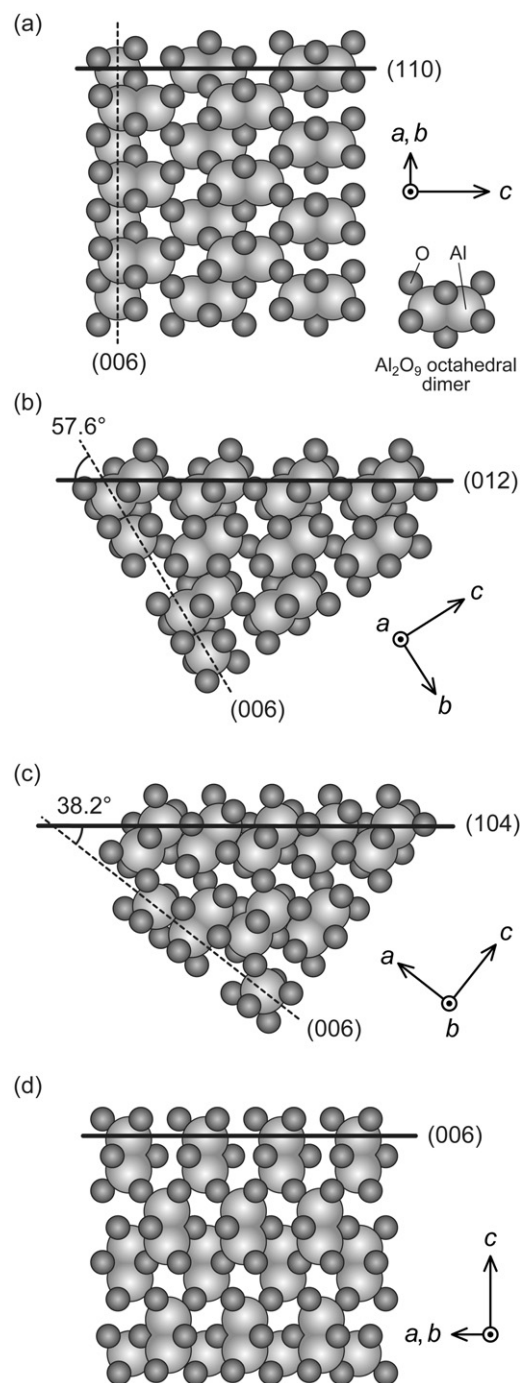


Fig. 10. Schematic of cross-sectional crystal structure of (006) (a), (104) (b), (012) (c) and (110) (d) planes of α - Al_2O_3 .

4. Conclusions

α - Al_2O_3 films were prepared by LCVD and the effects of deposition conditions on the phase, orientation and texture of Al_2O_3 film were investigated. At $P_{\text{tot}} = 0.93$ kPa, α - Al_2O_3 films were obtained in the region of $T_{\text{vap}} > 423$ K and $T_{\text{dep}} > 1100$ K. The orientation of α - Al_2O_3 film changed from (110) to (012) to (104) to (006) with increasing P_{tot} . Porous α - Al_2O_3 film was formed at high T_{vap} (443 K) and low P_{tot} (0.47 kPa). When T_{vap} decreased to 413 K, α - Al_2O_3 film had hexagonal and rectangular plate-like grains with finely faceted edges, while with increasing $P_{\text{tot}} = 0.93$ –1.4 kPa, (006)-oriented α - Al_2O_3 film with a hexagonal terrace texture was obtained.

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